Fuel Cell 3D Contour Simulation of Serpentine Flow Field Inside Cubic Channels

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Abstract. In this study, we described the behavior of the serpentine flow field inside the fuel of a proton exchange membrane fuel cell (PEMFC). Pressure drop, inlet velocity, oxygen and gas diffusion layer (GDL), and current density determine the design of the fuel cell. The other parameters predicted in the open literature, such as temperature, humidity, etc., are not taken into account here. COMSOL Multiphysics 5.4 Software was used to simulate and test the electrochemical reactions governing equations. Three inlet velocities and three of (1.5, 2.5, and 3.5 m/sec) cubic channels are used to show the effect of a turbulent flow without wall slip. By analyzing the results, it can be concluded that the serpentine flow field enhances the performance of fuel cells with cubic channel geometry. We found a good correlation between the current density and cell voltage at different inlet velocities.

Keywords: Serpentine Flow Field, 3D Contour Analysis, Fuel Cells, Cubic Channels.

1 Introduction

Fuel cells have recently attracted more researchers' attention. Applications in the industrial sector, due to global warming and fuel cost problems [1-5]. The last fifty years have seen many studies focused on compiling the use of water with another electrolyte to produce voltages and currents [6-9]. All these parameters played an important role in the design of the fuel cell membrane and the flow, whether single or two-phase [10-12]. The Serpentine flow is well known in the field of the fuel cell as a modern style of flow inside the channels. The designers developed new arts for the competitions of the fuel cell at serpentine flow. Recently published papers used the geometry of the channel as (M) [13]. In the catalyst layer and gas diffusion layer on the cathode side, they found a constant oxygen molar concentration. [14] uses a multiserpentine with slots and hybrid geometry. The best electrical performance was achieved with less pressure drop inside the geometry, and therefore, less noise was generated. Through a threedimensional wave flow channel, the channel geometry accelerates electrochemical reactions in the catalytic layer [15]. Their work results on the optimal flow channel are at minimum depth. The reaction inside the flow channel is observed more sensitive with relative humidity which showed by [16]. The experimental results shown at the increase in the humidity level, the performance of the cell increased.

In this study, we organized to preset the effect of the 3D serpentine flow field inside the three cubic channels of the fuel cell. The simulation is done used the free demo of the modern

software COMSOL Multiphysics 5.4. Most of the effective parameters were shown and compared with opened pieces of literature.

2 **Problem Formation**

The design for fuel cells depends upon the type of fuel utilized, current density, mass of the oxygen diffused and reacts with (GDL), and the pressure drop in each channel of the cell. The modern way is approached to use serpentine flow field and predicted with three parallel of equal lengths channels to an aligned of reactant distributed at fluid over the electrode area to identical in resistance flow. Nevertheless, this design may induce some gratuitously in the high-pressure drops with low-temperature fuel cells, obstructing because some of the water will be condensed which occurred in the bends. This design for serpentine channel to function used the cross-flow through channel-to-channel, to plot the essential convection in the porous material layer, and moderated it instead of a large cross-flow due to lead stagnant zones. The modern way is approached to use serpentine flow field and predicted with three parallel of equal lengths channels to an aligned of reactant distributed at fluid over the electrode area to identical in resistance flow.

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Fig 1. Model geometry.

Table 1. Shows all the input parameters to the simulated model.

Item	Expression	
Rib width	0.7 (mm)	
Channel width	0.8 (mm)	

Plate width	50 (mm)
Channel height	0.8 (mm)
GDL height	0.5 (mm)
Channel-to-channel space	1.5 (mm)
Channels Number	3
Channel inner radius at corners	0.25 (mm)
Cell Voltage (V)	0.4
Open circuit voltage (V)	1.23
Lumped anode + membrane resistance	0.285 (ohm*cm ²)
Cell temperature (K)	180+273.15
GDL porosity	0.4
Inlet mass fraction of H ₂ O (cathode)	0.023
Inlet mass fraction of oxygen (cathode)	0.228

2.1 Governing equations

The reactions occurring in the cell are described by the following formulas of the porous cathode reaction according to[13]:

$$O_2 + 4H^+ + 4e^- \leftrightarrow 2H_2O \tag{1}$$

Where, O₂ is oxygen, H hydrogen, H₂O is water.

The cumulative response in PEMFC fuel cell to calculate local current density, *iloc* which depend upon the oxygen concentrated according to the kinetic expression below[1]:

$$i_{loc} = i_0 \left(exp \left(\frac{\alpha_a}{RT} F \varphi_c \right) - \left(\frac{c_{0_2}}{c_{0_{2,ref}}} \right) exp \left(- \frac{\alpha_c}{RT} F \varphi_c \right) \right)$$
(2)

Where, R= universal gas constant, Temperature, α = transfer coefficient for cathode and anode, F= Faraday constant equal (96487 C mol⁻¹), φ = cathode overpotential, Local oxygen concentration and the reference concentration of oxygen.

The combinations for both boundaries of the electrolyte-electrode Interface set up the mass flow boundaries according to the stoichiometric coefficient of the electrodes reactant. The flow is assumed without wall slip used for the wall of the channel, while the wall slip conditions are considered for the GDL walls.

The cathode potential φ , is calculated by relating the various parts of the cell according to:

 $\varphi_{C} = E_{cell} - E_{OCV} - R_{iloc} \tag{3}$

Where E_{cell} is the cell potential, E_{OCV} is the open circuit cell voltage and R is the lumped resistance of membrane, anode, and the GDLs. the 3D flow of continuity equation is [1]:

$$\frac{\partial(\rho u)}{\partial x} + \frac{\partial(\rho v)}{\partial y} + \frac{\partial(\rho w)}{\partial z} = -\frac{\partial \rho}{\partial t}$$
(4)

Where, ρ =density, u, v,w are the velocity components of the x, y, and z respectively. In porous electrodes, the momentum equation can be written as:

$$\nabla \cdot \left(\epsilon \rho \underset{v}{\rightarrow} \underset{v}{\rightarrow} \right) = -\epsilon \nabla p + \nabla \epsilon \tau + Sm \tag{5}$$

The momentum equation in the x-direction is :

$$u \frac{\partial(\rho u)}{\partial x} + v \frac{\partial(\rho u)}{\partial y} + \omega \frac{\partial(\rho u)}{\partial z} = -\frac{\partial p}{\partial x} + \frac{\partial}{\partial x} \left(\frac{\mu \partial u}{\partial x}\right) + \frac{\partial}{\partial y} \left(\frac{\mu \partial u}{\partial y}\right) + \frac{\partial}{\partial z} \left(\frac{\mu \partial u}{\partial z}\right) + Sm$$
(6)

Where S_m is mass sink term. It can use the expression for y and z-direction (just changed the composition of each direction).

Diffusion mass flux vectored can calculate from Fick's equation is :

$$\xrightarrow{j}_{i} = \sum_{j=1}^{N-1} \rho \operatorname{Dij} \nabla \xrightarrow{j}_{yi}$$
(7)

Where Dij is the diffusivity of species.

The heat is transferred due to the chemical source, the Brinkman Equations is [17]:

$$\frac{1}{\epsilon_P}\rho(u,\nabla)u\frac{1}{\epsilon_P} = \nabla \left[-pI + K\right] - \left(\mu K^{-1} + \beta_F |u| + \frac{Q_m}{\epsilon_P^2}\right)u + F$$
(7)

$$\nabla (\rho u) = Q_m, \ K = \mu \frac{1}{\epsilon_P} (\nabla u + (\nabla u)^T - \frac{2}{3} \mu \frac{1}{\epsilon_P} (\nabla . u) I$$
(8)

Where k is the conductivity of the polar plate. And for Transport of Concentrated Species is:

$$\nabla . j_i + \rho(u. \nabla) \omega_i = R_j \tag{9}$$

$$j_i = -(\rho D_i^m \nabla \omega_i + \rho \omega_i D_i^m \frac{\nabla M_n}{M_n} - j_{c,i} + D_i^T \frac{\nabla T}{T}$$
(10)

$$D_i^m = \frac{1 - \omega_i}{\sum_{k \neq i} \frac{Xk}{D_i k}} , \quad M_n = (\sum_i \frac{\omega_i}{M_i})^{-1}$$
(11)

2.2 Physical Model

To run the model, the proton exchange membrane fuel cell (PEMFC) was designed and tested successfully by COMSOL Multiphysics 5.4 software. Fig 2 shows the three inlet cubic channels with three cubic outflow channels. All dimensions and the operating parameters that were input to the run are shown in Table1. The model test with three different inlet velocities ranged as (1.5, 2.5, and 3.5 m/sec). Table 2 shows the element quality of the fuel cell domain. Fig 3 and Fig 4 show the mesh of the whole fuel cell and the test of the mesh.



Fig 2. Dimensions of simulated.

Table 2. Statistical of element quality.

Description	Value
Min. element quality	0.08319
Average element quality	0.673
Tetrahedron	83113
Triangle	46997
Edge element	5041
Vertex element	140
Nu. of elements	251489
Element volume ratio	0.01344





Fig 3. 3D mesh domain test.



Fig 4. 3D snapshot for inlet and outlet meshing domain test.

3 Results and Discussions

3.1 Model Validation

Fig 5, explained the comparison of present work with previous work which was obtained from the experiment data's [1]. Their data compared at the ranged of the current density approximately from (0.22 to 1.07 A/cm²). Our results are simulated at the inlet velocity of (1.5 m/sec). The Figure illustrates the current is slight decreases with varying cell voltage. While for previous work, the clear slop of line increased the current density vs cell voltage.



Fig 5. The comparison between present work with [1].

3.2 Oxygen distribution

Fig 6 shows the oxygen distribution in the fuel cell channels in 3D. Oxygen concentration is higher at the inlet of each channel, and then gradually decreases towards the outlet. With opened pieces of literature, the reduction in oxygen concentration leads to a slower reaction rate of the cell. In the channels, the oxygen mass fraction is significantly higher. As inlet velocity increases,

the high concentration value of oxygen starvation is remarkably reduced at the cell's outlet. In Figure 6, the oxygen distribution is shown at an inlet velocity of (1.5 m/sec) while Figs 6-b and 6-c show inlet velocities of 2.5 and 3.5 m/sec.



Fig 6. Distribution of oxygen concentration.

3.3 Velocity Distribution

Fig 7 shows the 3D velocity distribution of the water inside the fuel cell channels. Each cell of the channel displays a serpentine flow field. The distribution inside the channel changed as velocity increased at the inlet. The more reaction between gas and diffusion layer, the more reliable the reaction. In addition to the velocity distribution, the pressure distribution can also be used to show the behavior inside the fuel channel. As inlet velocity increases, the pressure increases and gradually decreases until it reaches a lower value at the outlet as shown in Fig 8.



Fig 7. 3D velocity distribution at three inlet velocities.

3.4 Current Density

Fig 9 illustrates the current density distribution at the cathode base. The most uniform distribution of current density is seen at serpentine flow fields, as is the distribution of oxygen molar concentration. Gas diffusive layers in the cathode release more reactants and can participate in the electrochemical reaction of the catalyst layer, as shown in Fig 10. These results can be used to improve and implement serpentine flows in fuel cell PEMFCs.



Figure 8. 3D pressure drop distribution at three inlet velocities.



Fig 9. 3D current density distribution at the cathode at three inlet velocities.



Fig 10. 3D gas diffusion layer (GDL) distribution at the cathode at three inlet velocities.

4 Conclusion

Three different inlet velocities were used to model the serpentine flow field in 3D. We use three cubic channels with inlet and outlet dimensions of (0.8*0.8 mm with average plate width of 50mm), all other parameters are listed in Table 1. By using the COMSOL Multiphasic 5.4 software (Free Demo), the model of the simulation is successfully run to test the effects of varying the inlet velocity on the performance of the fuel cell. The following conclusion follows:

- 1. The higher performance to produce the current density with serpentine flow field arrangement is obtained in the present work and so for observed by the former researchers.
- 2. 3D oxygen distribution uniformly at higher values inlet and gradually reduction it to outlet due to decreases the mass fraction when it reacts with GDL.
- 3. 3D current density distributed at the cathode is observed that the most uniform at serpentine flow field. The gas diffusive layer in the cathode is more reactant given participated the catalyst interchanged.
- 4. The increases in inlet velocities that lead to high-pressure drop values were observed through the inlet and outlet of each channel.

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